

SESQUITERPENE LACTOLS FROM *LASERPITIUM HALLERI*

GIOVANNI APPENDINO, MARIA GRAZIA VALLE and PIERLUIGI GARIBOLDI*

Istituto di Chimica Farmaceutica e Tossicologica, Facoltà di Farmacia, C.so Raffaello 31, 10125 Torino, Italy; *Dipartimento di Chimica Organica ed Industriale, Università di Milano, via Venezian 21, 20133 Milano, Italy

(Received 8 September 1986)

Key Word Index—*Laserpitium halleri*; Umbelliferae; sesquiterpene lactols.

Abstract—Three new pairs of anomeric sesquiterpene lactols have been isolated from the roots of *Laserpitium halleri*. They were obtained in pure form only after methylation of the hemiacetalic hydroxyl group, which resulted in the exclusive formation of only one methyl ketal from each pair of anomers. The stereochemistry at the ketalic centre of the methyl derivatives was assessed by correlation with methylhallerin, a compound of known configuration.

INTRODUCTION

As a part of an investigation on alpine plants, we reported the isolation of hallerin (**1a, b**), a mixture of anomeric sesquiterpene lactols, from the roots of *Laserpitium halleri* Crantz subsp. *halleri* [1].

Inspection of the high-field ¹H NMR spectra of crude samples of hallerin showed that, besides the expected resonances for **1a, b**, three other sets of split signals were present, accounting for ca 10% of the material, and most probably representing other pairs of anomeric lactols. We present here the structural elucidation of these minor products.

RESULTS AND DISCUSSION

The separation of the mixture of lactols was simplified by treatment with methyl iodide and silver(I) oxide in dry dichloromethane, since exclusively one methylketal was formed from each pair of anomers. Furthermore, the methyl derivative of hallerin (**1c**), representing the most abundant constituent of the mixture, is a highly crystalline compound, and most of it could be removed by crystallization from hexane. By column chromatography of the mother liquors, it was thus possible to obtain pure **2c**, and a mixture of **3c** and **4c**, which could be further separated on AgNO₃-coated silica gel.

Compound **2c** is the methyl derivative of a mixture of anomeric elemenes (isohallerin **2a, b**), previously prepared by Cope rearrangement of hallerin [1], while **3c** and **4c** are the methyl derivatives of anomeric eudesmane lactols (**3a, b; 4a, b**).

The stereochemistry at the ketalic centre of methylhallerin (**1c**) was established as *S* (β -OMe) by an X-ray diffraction study [2]. That of the other methyl ketals was assessed by correlation with methylhallerin [1]. Cope rearrangement of **1c** gave **2c**, while transannular cyclization of **1c** under controlled conditions (see Experimental) afforded **3c** and **4c**, showing that the stereochemistry at the ketalic centre is the same in all these compounds.

Comparison of the ¹H NMR spectra of **1c** and **2c-4c** shows that Cope rearrangement to **2c** and transannular

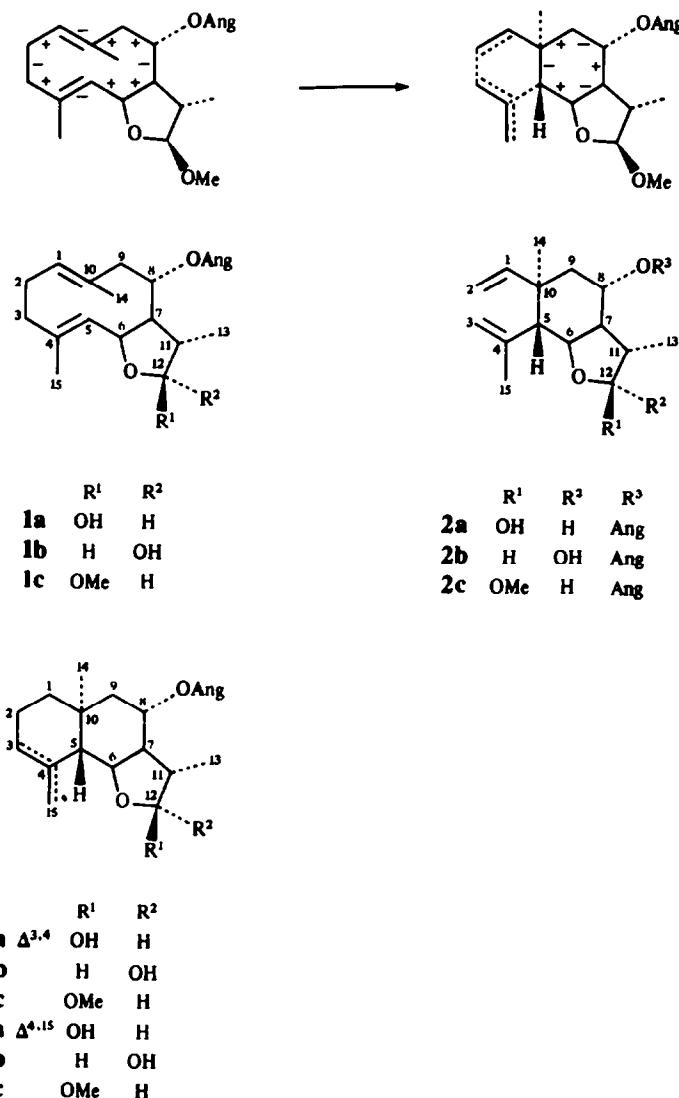
Table 1. ¹³C NMR data for compounds **1c**, **2a**, **2b**, **2c** (50 MHz, CDCl₃, TMS internal standard)

C	1c	2a	2b	2c
1	129.17 <i>s</i>	147.46 <i>d</i>	147.90 <i>d</i>	147.99 <i>d</i>
2	25.88 <i>t</i>	114.32 <i>t</i>	113.87 <i>t</i>	113.73 <i>t</i>
3	39.63 <i>t</i>	110.80 <i>t</i>	110.69 <i>t</i>	110.66 <i>t</i>
4	134.07 <i>s</i> *	143.20 <i>s</i>	143.84 <i>s</i>	143.84 <i>s</i>
5	127.33 <i>d</i>	46.76 <i>d</i>	50.44 <i>d</i>	50.40 <i>d</i>
6	76.30 <i>d</i>	77.43 <i>d</i>	78.15 <i>d</i>	78.48 <i>d</i>
7	53.32 <i>d</i>	53.75 <i>d</i>	54.13 <i>d</i>	53.35 <i>d</i>
8	72.05 <i>d</i>	69.77 <i>d</i>	70.64 <i>d</i>	69.93 <i>d</i>
9	44.88 <i>t</i>	41.04 <i>t</i>	43.08 <i>t</i>	38.61 <i>t</i>
10	133.95 <i>s</i> *	39.92 <i>s</i>	40.32 <i>s</i>	40.21 <i>s</i>
11	42.33 <i>d</i>	41.04 <i>d</i>	43.08 <i>d</i>	42.95 <i>d</i>
12	109.71 <i>d</i>	98.48 <i>d</i>	104.87 <i>d</i>	19.76 <i>d</i>
13	17.90 <i>q</i> †	15.72 <i>q</i>	11.94 <i>q</i>	16.47 <i>q</i>
14	16.78 <i>q</i>	20.61 <i>q</i>	19.35 <i>q</i>	19.44 <i>q</i>
15	21.00 <i>q</i> †	25.35 <i>q</i>	24.49 <i>q</i>	25.16 <i>q</i>
OMe	54.46 <i>q</i>			56.58 <i>q</i>

The ¹³C signals for the angelate group were almost identical in **1c**, **2a**, **2b** and **2c**. Those observed in **1c** are given as representative: C-1' 166.71 *s*, C-2' 128.64 *s*, C-3' 136.44 *d*, C-4' 15.52 *q*, C-5' 20.52 *q*.

* † Interchangeable signals.

cyclization to **4c** are accompanied by inversion of the fragment C-6-C-10, which results in flipping from a boat to a chair conformation of the cyclohexane ring bound to the γ -lactol ring. These changes are outlined in Fig. 1 according to the torsion angle notation [3]. Interestingly, the conformation of ring B is different in the eudesmane **4c** and **3c**, as shown by the different splitting pattern of H-8 (*q*, *J* = 7.0 Hz in **3c**; *br s*, in **4c**). Although it is difficult to derive the geometry of ring B in **3c** from simple inspection of the coupling constants around the ring, the pattern is similar to that reported for isosilerolide [4], in which ring B adopts a twist-boat conformation, at least in the solid state [5]. These differences are difficult to rationalize, and



Scheme 1. Top, conformational changes during Cope rearrangement and transannular cyclization of 1c.

were not observed in α - and β -cyclohallerin lactones, the 1-hydroxylated γ -lactones corresponding to 3c and 4c [1].

The stereochemical outcome of the methylation of hallerin and related mixtures of anomeric lactols has been discussed elsewhere [2]. It is interesting to note that substitution of dichloromethane for DMF allows the selective methylation of the more acidic hemiacetalic hydroxyl in the presence of alcoholic hydroxyls [1].

It has been reported that under certain conditions silver(I) ions can catalyse the interconversion of elemene and eudesmane derivatives possibly via the corresponding germacrene derivatives [6]. However no reaction of this type was observed by treatment of pure 1c and 2c with Ag_2O in dichloromethane.

Lactols 1a, b-4a, b are structurally related to lactones isolated from the Laserpitieae tribe of the Umbelliferae from which they differ, apart the ester moiety, only in the absence of an oxygenated function at C-11. It is not clear, however, if sesquiterpene lactols in the Laserpitieae tribe are involved in the biogenesis of sesquiterpene lactones, of which they might represent the ultimate

precursors. Sesquiterpene lactones, were not isolated from the roots of *Laserpitium halleri*, whereas investigation of the fruits gave only sesquiterpenoids with an unoxidized C-7 side-chain [7].

EXPERIMENTAL

Silica gel 60 (70–230 mesh, Merck) was used for CC; AgNO_3 -coated silica gel was prepared by suspending silica gel (TLC-grade, 100 g) in a 5% soln of AgNO_3 in MeCN (200 ml), and removing the solvent under vacuum. The material prepared in this way was activated before use by heating at 120° for 2 hr. Crude hallerin was obtained from the roots of *Laserpitium halleri* as described in ref. [1].

Methylation of the mixture of lactols. A 2.0 g sample of crude hallerin (6.2 mmol) was dissolved in 25 ml CH_2Cl_2 (distilled from P_2O_5) and 1.60 g Ag_2O was added, followed by an excess (2 ml, 39 mmol) of MeI. The suspension was stirred at room temp. for 6 hr, and then filtered and evapd. A solid material was obtained, which was crystallized from hexane to give pure methylhallerin (1c, 1.40 g). The mother liquors were chromatographed on silica

gel (20 g, hexane-EtOAc, 6:1) to give pure 2c (160 mg), an additional 200 mg of 1c and 95 mg of a mixture of 3c and 4c, which was further separated by CC on silver-coated silica gel (20 g, hexane-EtOAc, 9:1) to give 20 mg 3c and 32 mg 4c.

Methylhallerin (1c). Leaflets from hexane, mp 91°, $[\alpha]_D^{25} - 9$ (CHCl₃; *c* 0.95); IR $\nu_{\text{max}}^{\text{KBr}}$ cm⁻¹: 2900, 1720, 1660, 1460, 1390, 1240 and 1160; EI-MS 70 eV, *m/z* (rel. int.): 348 [M]⁺ (C₂₁H₃₂O₄) (3), 83 [C₅H₉O]⁺ (100); ¹H NMR (CDCl₃, 200 MHz, TMS as internal standard) [the resonance values of the protons of the angelate were constant in all the compounds reported here: those for 1c are given as representative]: δ 5.98 (1H, *qq*, *J* = 7.0, 1.5 Hz, H-3'), 1.97 (3H, *dq*, *J* = 7.0, 1.5 Hz, H-4'), 1.89 (3H, *q*, *J* = 1.5 Hz); δ 5.26 (1H, *dq*, *J* = 10.4, 9.8, 5.0 Hz, H-8), 5.15 (1H, *tq*, *J* = 8.0, 8.0, 1.2 Hz, H-1), 4.92 (2H, *m*, H-5, H-6 overlapped signals), 4.48 (1H, *d*, *J* = 1.3 Hz, H-12), 3.23 (3H, *s*, -OMe), 2.85 (1H, *ddd*, *J* = 5.0, 12, 1.5 Hz, H-9), 2.65 (1H, *qt*, *J* = 7.2, 1.3, 1.3 Hz, H-11), 1.67 (3H, *d*, *J* = 1.5 Hz, H-14), 1.43 (3H, *d*, *J* = 1.5 Hz, H-15), 1.13 (3H, *d*, *J* = 7.2 Hz, H-13).

Methylisohallerin (2c). Colourless oil, $[\alpha]_D^{25} + 30$ (CHCl₃; *c* 1.83); IR $\nu_{\text{max}}^{\text{CHCl}_3}$ cm⁻¹: 1720, 1640, 1460, 1390 and 1230; EI-MS 70 eV, *m/z* (rel. int.): 348 [M]⁺ (C₂₁H₃₂O₄) (6), 83 [C₅H₉O]⁺ (100); ¹H NMR (CDCl₃, 200 MHz, TMS as internal standard): δ 5.74 (1H, *dd*, *J* = 17.0, 11.0 Hz, H-1), 5.22 (1H, *td*, *J* = 4.0, 4.0, 2.2 Hz, H-8), 5.02 (1H, *m*, H-3a), 4.90 (1H, *dd*, *J* = 11.0, 1.0 Hz, H-2a), 4.86 (1H, *dd*, *J* = 17.0, 1.0 Hz, H-2b), 4.74 (1H, *br s*, H-3b), 4.57 (1H, *d*, *J* = 4.0 Hz, H-12), 4.43 (1H, *dd*, *J* = 11.0, 6.5 Hz, H-6), 3.68 (3H, *s*, -OMe); 2.33 (1H, *d*, *J* = 11.0 Hz, H-5); *ca* 2.12 (1H, *m*, H-11), 1.78 (3H, *br s*, H-15); 1.57 (1H, *dd*, *J* = 16.0, 4.0 Hz, H-9a), 1.21 (3H, *d*, *J* = 6.4 Hz, H-13); 1.08 (3H, *s*, H-14).

Methyl- α -cyclohallerin (3c). Gum, $[\alpha]_D^{25} - 140$ (CHCl₃; *c* 0.51); IR $\nu_{\text{max}}^{\text{CHCl}_3}$ cm⁻¹: 1720, 1460, 1390, 1240-1220; EI-MS 70 eV, *m/z* (rel. int.): 348 [M]⁺ (C₂₁H₃₂O₄) (4), 83 [C₅H₉O]⁺ (100); ¹H NMR (CDCl₃, 200 MHz, TMS as internal standard): δ 5.40 (1H, *br s*, H-3); 5.10 (1H, *q*, *J* = 7.0 Hz, H-8), 4.80 (1H, *d*, *J* = 5.0 Hz, H-12), 4.18 (1H, *dd*, *J* = 10.5 and 7.5 Hz, H-6); 3.36 (3H, *s*, -OMe); 1.83 (3H, *br s*, H-15); 1.12 (3H, *d*, *J* = 6.2 Hz, H-13); 0.96 (3H, *s*, H-14).

Methyl- β -cyclohallerin (4c). Colourless oil, $[\alpha]_D^{25} + 8$ (CHCl₃; *c* 0.60); IR $\nu_{\text{max}}^{\text{CHCl}_3}$ cm⁻¹: 1720, 1650, 1450, 1390, 1240-1210; EI-MS 70 eV, *m/z* (rel. int.): 348 [M]⁺ (C₂₁H₃₂O₄) (3), 83 [C₅H₉O]⁺ (100); ¹H NMR (CDCl₃, 200 MHz, TMS as internal standard): δ 5.14 (1H, *br s*, H-8), 4.94 (1H, *br s*, H-15a), 4.83 (1H, *br s*, H-15b), 4.62 (1H, *d*, *J* = 3.0 Hz, H-12); 4.45 (1H, *dd*, *J* = 10.5 and 6.0 Hz, H-6), 3.30 (3H, *s*, -OMe); 1.24 (3H, *d*, *J* = 6.2 Hz, H-13); 0.88 (3H, *s*, H-14).

Cope rearrangement of methylhallerin (1c). A 1.10 g sample of 1c was heated under vacuum (water pump) at 180° in a Kugelrohr apparatus for 5 min. A yellowish oil was obtained, which was filtered on a short column of silica gel (4 g) to give 996 mg of 2c as a colourless oil, having spectral data (¹H NMR, IR) identical to those of 2c obtained from the methylation of the mixture of lactols from *L. halleri*.

Transannular cyclization of methylhallerin (1c). A 100 mg sample of 1c was dissolved in 0.5 ml of CDCl₃ in an NMR tube, and 5 mg of *p*-toluenesulphonic acid were added. The course of the reaction was followed by ¹H NMR (60 MHz). After 10 min. all of 1c had reacted, and the reaction was quenched by addition of 0.7 ml of a saturated NaHCO₃ soln. The contents of the tube was poured into CH₂Cl₂ (10 ml), and the tube was rinsed \times 3 with the same solvent. After separation of the phases and drying (MgSO₄), 91 mg of a yellow oil were obtained. After 2 separations on silver-coated silica gel (hexane-EtOAc 9:1 as eluent), 12 mg of 3c and 21 mg of 4c were obtained, identical (¹H NMR, 200 MHz) with the eudesmanes obtained from the methylation of the mixture of lactols from *L. halleri*.

Acknowledgements—This work was supported in part by the Ministero della Pubblica Istruzione (Fondi 60 and 40%) and by the Assessorato Agricoltura e Foreste, Servizi Forestali della Regione Autonoma Valle d'Aosta. We are grateful to Prof. G. M. Nano for his encouragement and useful advice. We thank Dr. C. Bicchi and Dr. C. Frattini for the mass spectra, and Dr. L. Poggio, Giardino Botanico Alpino Paradisia, Valnontey, Cogne, for the plant material.

REFERENCES

1. Appendino, G. and Gariboldi, P. (1983) *J. Chem. Soc. Perkin Trans 1*, 2017.
2. Appendino, G., Chiari, G., Viterbo, D. and Ugliengo, P. (1987) *J. Chem. Soc. Perkin Trans 2* (in press).
3. Toromanoff, E. (1980) *Tetrahedron* **36**, 2809.
4. Holub, M., Buděšínský, M., Smitalová, Z., Saman, D. and Rychlewska, V. (1982) *Tetrahedron Letters* 4853.
5. Rychlewska, V. (1983) *J. Chem. Soc. Perkin Trans 2*, 1675.
6. Jain, T. C. and McCloskey, J. E. (1972) *Tetrahedron Letters* 5139.
7. Appendino, G., Valle, M. G., Gariboldi, P. (1986) *J. Chem. Soc. Perkin Trans 1*, 1363.